



Course description:

This course provides the student with the principles of instrumental analysis upon which modern measuring devices are based. This includes the exploration of the instrumental methods of analysis used to check the purity of raw material and quality control of pharmaceutical preparation; using chromatographic methods, spectroscopic methods; UV-Visible, IR, NMR and Mass spectroscopy.

Aims of the course:

1. To understand the principle of the instrumental methods of analysis including spectroscopic methods; UV-Visible, IR, NMR and Mass spectroscopy.
2. To learn how to interpret the result obtained from the different spectroscopic methods.

Intended Learning Outcomes: (ILOs)

A. Knowledge and Understanding

A1. Concepts and Theories:

Be introduced to a wide range of techniques of modern analytical chemistry.

A2. Contemporary Trends, Problems and Research:

Identification of chemical structures using IR, NMR and MS.

B. Subject-specific skills

B1. Problem solving skills:

To study the UV, Visible and to solve problems using Beer's-Lambert's law.
Identification of unknown chemical structures using IR, NMR and MS.

B2. Application of Methods and Tools:

To apply instrumental methods in drugs identification and quantification.

C. Critical-Thinking Skills

C1. Analytic skills: Assess

To analyze wide range of drugs qualitatively and quantitatively.

C2. Strategic Thinking:

To design plans and strategies to analyze different compounds within a mixture.

C3.Creative thinking and innovation:

Students should correlate instruments with drug analysis, dose preparation, pharmacokinetic calculations, and plants extracts analysis.

D. General and Transferable Skills (other skills relevant to employability and personal development)**D1. Communication:**

Student should work in groups and gain communication skills to solve different problems.

D2.Teamwork and Leadership:

Group work and tasks division between group members especially when NMR data need to be analyzed.

Teaching strategies in course outline:

Teaching strategies in course outline	
✓ Lectures	<input type="checkbox"/> Field visits
<input type="checkbox"/> Online Lectures	<input type="checkbox"/> Role play
<input type="checkbox"/> Practical training / laboratory	<input type="checkbox"/> Presentation
<input type="checkbox"/> Seminar / Workshop	✓ Quizzes
✓ Moodle	<input type="checkbox"/> Online resources (Kahoot., etc)
<input type="checkbox"/> Case study	✓ Team based learning
<input type="checkbox"/> Flipped classroom	<input type="checkbox"/> Social related activities (chat)
<input type="checkbox"/> Simulation	✓ Other assignments and homework
<input type="checkbox"/> Others :	

Competencies:

Competencies
(Learner)
3.1 (problem solver)
4.1 (Self- aware)

Course structures:

Week	ILOs	Topics	Teaching	Assessment
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			Procedure	methods
1	A1,B1	Introduction to Instrumental analysis		
2+3+4	A1,B1,C1,D1	UV-Visible Spectroscopy	Lectures + Classroom discussions	Quiz
5+6+7	A1,A2,B1,D1,D2	Infrared Spectroscopy (IR)	Brain storming exercises + Classroom discussions	Classroom worksheets
8+9+10	A1,A2,B1,D1,D2	Proton Nuclear Magnetic Resonance (^1H NMR) + ^{13}C NMR	Lectures + Brain storming exercises	Classroom worksheets
11+12	A1,B2,C1,C2,C3,D1	Chromatographic Theory and High Performance Liquid Chromatography (HPLC)	Animation + Lectures + Classroom discussions	Classroom worksheets
13+14	A1,B2,C1,C2,C3,D1	Gas Chromatography (GC)	Lectures	Classroom worksheets
15		Final Exam		

References:

A. Main Textbook:

Principles of Instrumental Analysis - 6th Edition by Douglas A. Skoog, F. James Holler, and Stanley Crouch

B. Supplementary Textbook(s):

Pharmaceutical Analysis, A Textbook for Pharmacy Students and Pharmaceutical Chemists, David G. Watson; 2nd edition (2005).

Undergraduate Instrumental Analysis, James W. Robinson, Eileen M. Skelly Frame; 6 edition (2004)

Assessment Methods:

Assessment	Grade
- First Exam	20%
- Second Exam	20%
- Assignments (reports, quizzes, homeworks, participation)	10%
- Final Examination	50%